

## Small Scale Lab 25 Answers

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### answer key small scale lab

Chapter 11 Small-Scale Lab Section 11.3 Precipitation Reactions: Formation of Solids, page 345 Analysis 1. a. Na 2CO 3 2AgNO 3 → 2NaNO 3 3Ag 2CO 3(s) b. 2Na 3PO 4 3Pb(NO 3) 2 → 6NaNO 3 Pb 3(PO 4) 2(s) 2. Sodium hydroxide reacts with calcium chloride to form sodium chloride and solid calcium hydroxide. 3. Mixings d,n, and o all gave no visible

### Chapter 11 Small-Scale Lab

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Chemistry Small-Scale Laboratory Manual v Small-Scale Laboratory Techniques Small-scale chemistry uses smaller amounts of chemicals than do other chemistry methods. The hazards of glass have been minimized by the use of plastic labware. If a chemical reaction must be heated, hot water will provide the needed heat. Open flames or burners are seldom

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My students enjoy working with the small-scale lab activities. For the most part, the 41 labs use fewer materials and very small amounts of reactants (hence the name "Small-Scale"). Set-up time is usually less than setting up for the full lab activities in the related Chemistry Lab Manual. Prentice Hall Chemistry: Small Scale Chemistry ...

### Pearson Small Scale Chemistry Laboratory Manual Answers

Used for accurately measuring and delivering small volumes of liquid-usually 0.1-10 mL. ring clamp Attached to ring stand and with wire gauze used to hold beakers or flasks while they are heated by a gas burner.