

# Nine Solutions Lab Answers

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### **Nine Solutions Lab Answers**

During this part of the experiment, you will be given a set of nine solutions labeled "A" through "I". They contain the same nine solutions as in Part I. However they are not identified by number. For example "A" could be any one of the numbered solutions. Each set contains the same solutions; however the labeling is unique for each set.

### **NINE-SOLUTION PROBLEM**

(Revised Fall 2009) Chemistry 161 Lab - K. Marr Green River Community College Lab 5 - Page 1 of 6  
Lab 5. The Nine-Solution Problem Prelab Assignment Before coming to lab: Use the handout "Lab Notebook Policy" as a guide to complete the following sections of your report for this lab exercise before attending lab: Title and Date of Lab, Introduction, Materials/Methods and Data Tables (see page 3 ...

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### **Lab 5\_The Nine Solution Problem\_F2009 - (Revised Fall 2009 ...**

Purpose The purpose of this lab is to use the characteristics of different reactions with aqueous mixtures to identify the solutions that are being used for the lab (Radhakrishnan, 2016). In order to conduct this experiment, each of the 36 binary combinations from the nine solutions were mixed together and the reactions that occurred were noted down in order to single out each solution based ...

### **The Problem of 9 Unknown Solutions - The Problem of 9 ...**

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When upon mixing the aqueous solutions we will classify the nine reagents in solutions based off how they react with each other in the diagram given to us in the lab manual. With our given unknown (#8) we should be able to identify the reagent solution by looking at our previous combination results and discover that our unknown is one of the ...

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### **Lab Report 4 - CHEM 1100 General Chemistry I - CSULA - StuDocu**

Nine Bottles Experiment By: Jenna, Bailey, and Claire Second weeks lab beaker #37 First weeks data Methodology Unknown #4 and #6  $\text{NaOH} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgOH}$  Unknown #1 :  $\text{H}_2\text{SO}_4$  Unknown #2 :  $\text{NaCl}$  Unknown #3:  $\text{NaBr}$  Unknown #4:  $\text{AgNO}_3$  Unknown #5:  $\text{KI}$  Unknown #6:  $\text{NaOH}$  Unknown #7:

### **Nine Bottles Experiment by jenna boyle - Prezi**

I am given 9 bottles with solutions in each bottle that remain nameless. The object of this is to figure out which bottles contain the following solutions below... Solutions: barium chloride....

### **Doing the NINE BOTTLE PROBLEM for chem lab ... - Yahoo Answers**

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## **NCERT Solutions Class 9 Social Science**

Ok, so I'm sure many of you have heard of this lab. The instructor hands out 9 unknown solutions, and it's your goal to label each. The names have been given to us a day before and we are allowed to research the elements. The list of elements include:  $\text{BaCl}_2$ ;  $\text{Cu}(\text{NO}_3)_2$ ;  $\text{CuSO}_4$ ;  $\text{HCl}$ ;  $\text{Hg}_2(\text{NO}_3)_2$ ...

## **9 Solution Problem Lab | Physics Forums**

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## **NCERT Class 9 Science Lab Manual - CBSE Tuts**

\*\*\*Identifying 9 unknown solution lab: You will be given 9 test tubes containing ~2 mL of your unknown solutions. You know only that they are aqueous solutions of the following compounds:  $\text{ZBr}$ ,  $\text{ZOH}$ ,  $\text{Z}_2\text{S}$ ,  $\text{Z}_2\text{SO}_4$ ,  $\text{ZClO}_3$ ,  $\text{ZIO}_3$ ,  $\text{NH}_4\text{Cl}$ ,  $\text{BaCl}_2$ , and  $\text{Pb}(\text{NO}_3)_2$ , where Z is  $\text{Na}^+$ ,  $\text{K}^+$  or  $\text{Li}^+$ .  
1. DEVELOP A STRATEGY THAT INVOLVES LOGICAL STEPS TO IDENTIFY ALL NINE OF YOUR SOLUTIONS.

## **\*\*\*Identifying 9 Unknown Solution Lab: You Will Be ...**

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The Nine-Bottle Challenge Name\_\_\_\_\_ Background: Almost all salts ("salt" is a name for an ionic compound) exist in water solutions as separated ions. For example, a solution of sodium chloride contains  $\text{Na}^+$  ions and  $\text{Cl}^-$  ions, and no  $\text{NaCl}$  species. When solutions of different salts are mixed, the ionic species present may

### **The Nine-Bottle Challenge**

Match solutions using simple chemical tests. This video is part of the Flinn Scientific Best Practices for Teaching Chemistry Video Series, a collection of over 125 hours of free professional ...

### **Mystery Solutions Lab**

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